

Control and Prediction of Molecular (& Crystal Properties by Multilevel Strategies

Gerit Brandenburg <g.brandenburg@ucl.ac.uk> | 26th of February 2018

VORTRAG IM RAHMEN DES BERUFUNGSVERFAHRENS 530, TECHNISCHE UNIVERSITÄT DARMSTADT, GERMANY

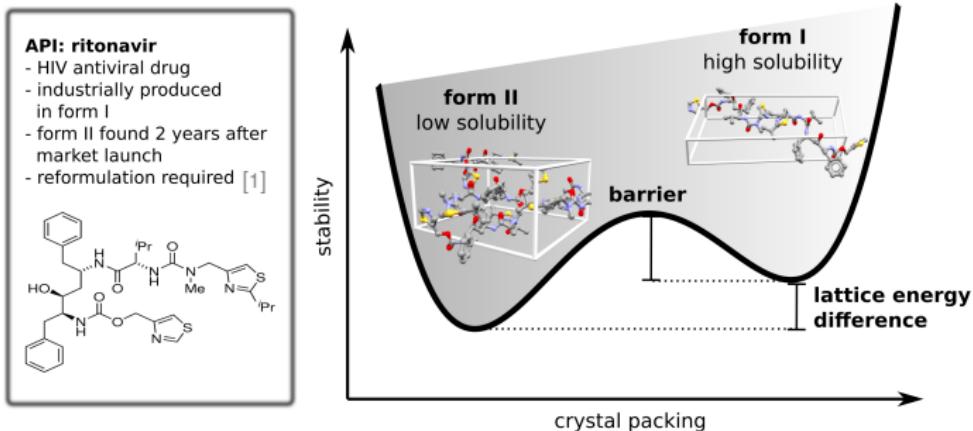


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Outline of talk

- 1 Introduction
- 2 Simulation based crystal structure prediction
- 3 Frustrated Lewis pair chemistry in the solid state
- 4 Quantum Monte-Carlo for molecular materials
- 5 SCAN-D3: Boosting the accuracy of meta-GGAs
- 6 Project outlook

Academic and industrial interest in molecular materials

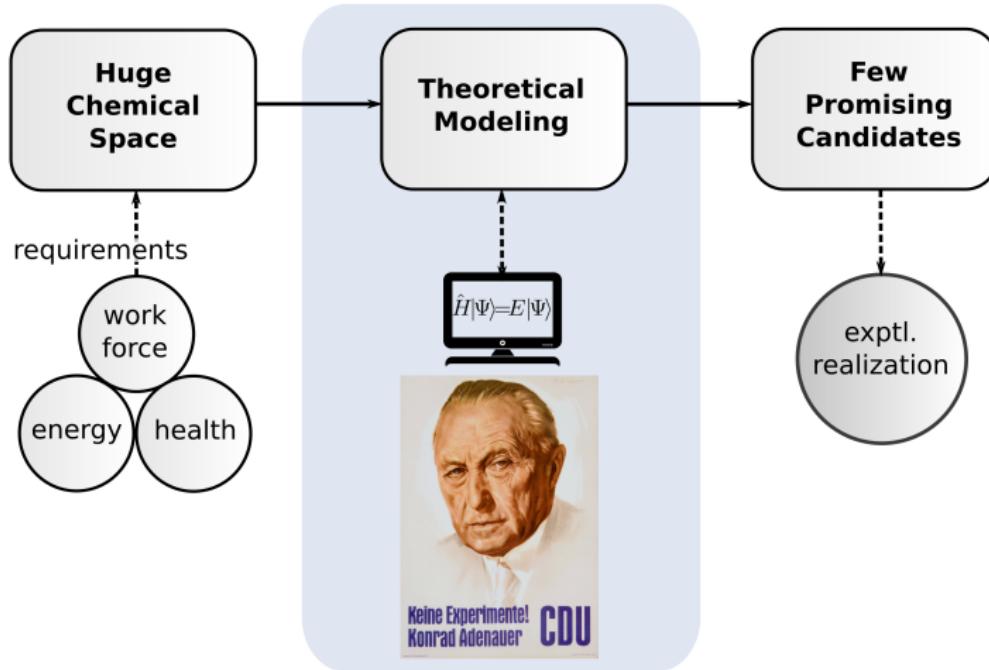


- tools to predict possible polymorphs would be valuable^[2]
→ Simulation methods aim at complementing experimental screening

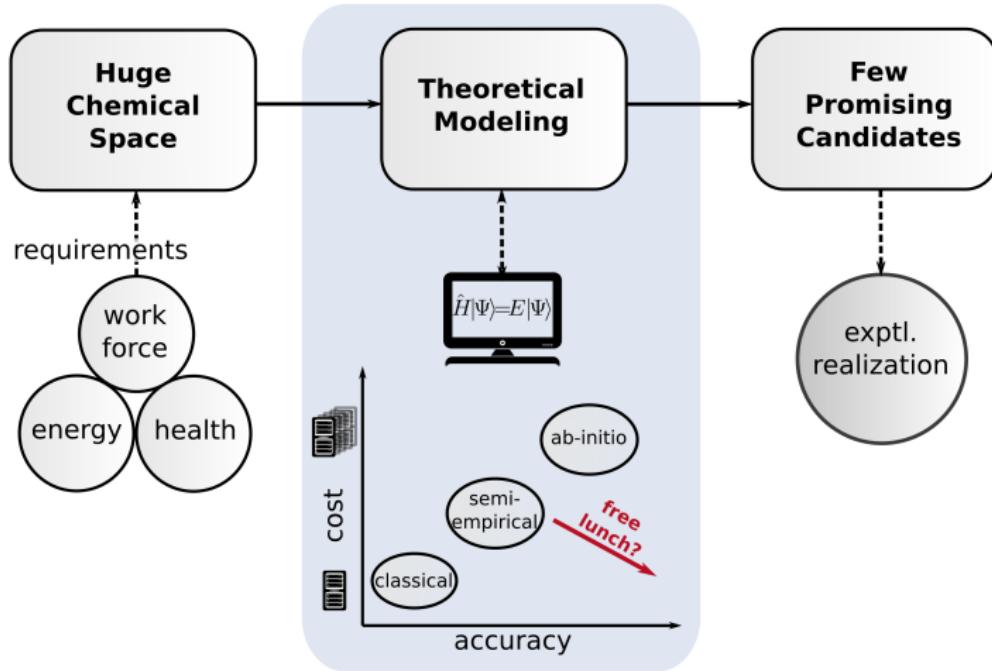
[1] J. Bauer, et al., *J. Pharm. Res.* 18, 859-866 (2001).

[2] S. L. Price, *JGB, Molecular Crystal Structure Prediction*; Elsevier Australia ISBN: 9780128098356 (2017).

Chemical and material discovery can employ computational models



Chemical and material discovery can employ computational models



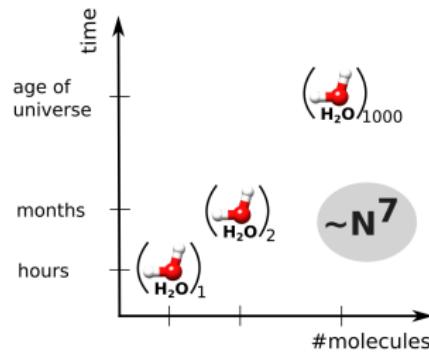
Exact simulation of extended systems computationally very demanding

Paul Adrian Maurice
Dirac (1902-1984)



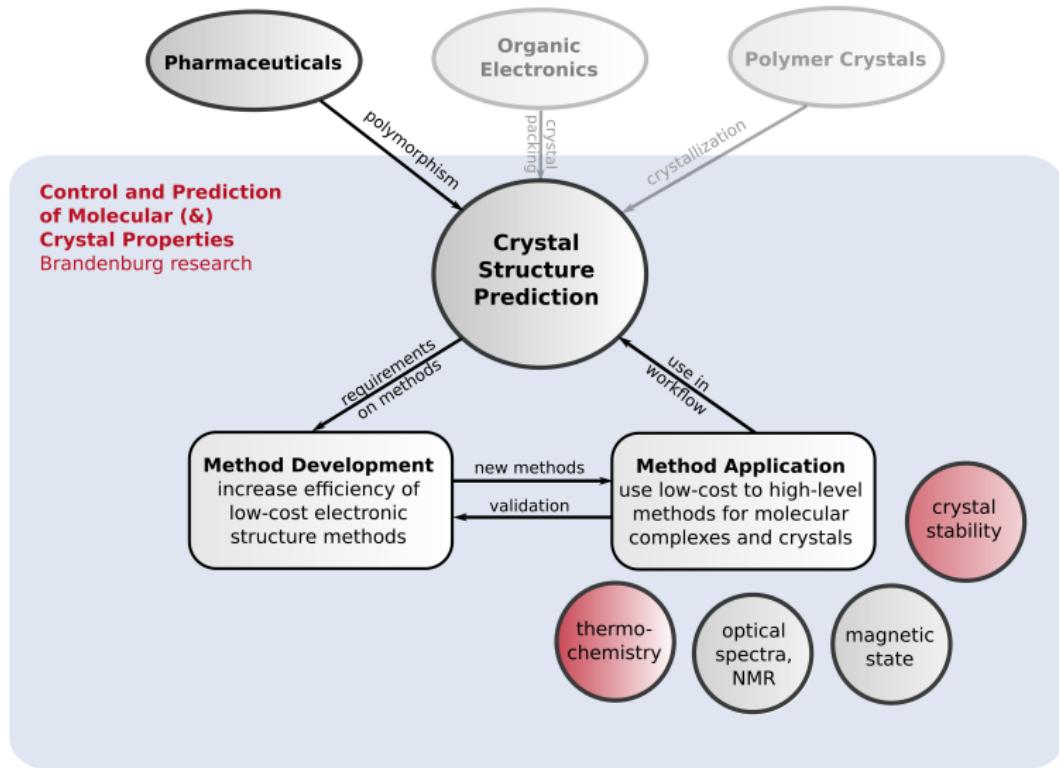
"The underlying physical laws (...) of a large part of physics and the **whole of chemistry** are thus completely known,

and the difficulty is only that the exact application of these laws leads to equations much too complicated to be soluble"^[1]



^[3]P. A. M. Dirac, Proc. Roy. Soc. Ser. A 123, 714 (1929).

Brandenburg research concept



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Predict most stable crystal polymorphs based on the molecular diagram



The inability to predict something as simple as how a molecule would crystallize is one of the continuing scandals in the physical sciences.^[4–6]

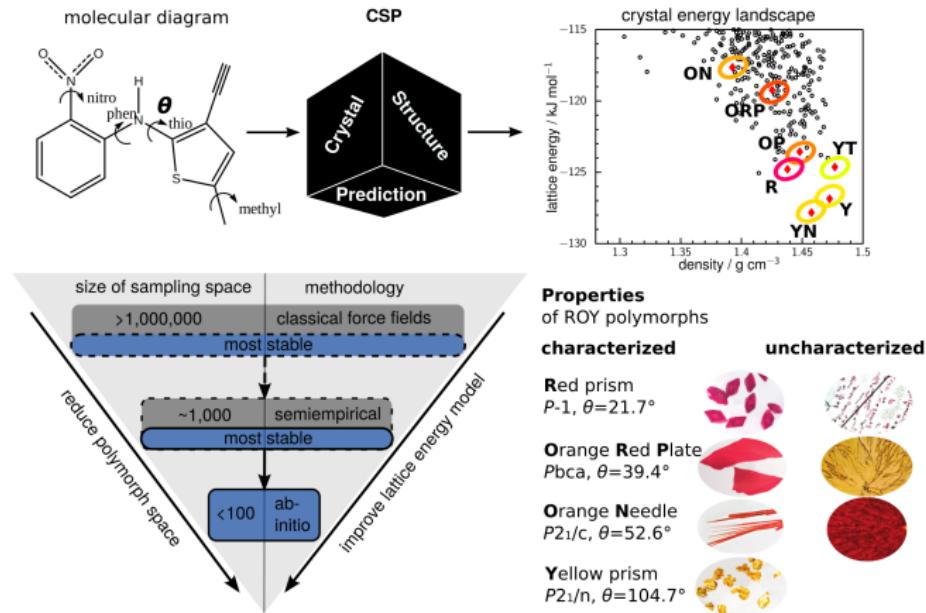
Task

- molecule is chosen due to its chemical/physical/biological properties
- based on the molecular diagram only, the most stable crystal structures should be predicted
- predict properties of interest for the most promising candidates

^[4] A. Gavezzotti, *Acc. Chem. Res.* **27**, 309-314 (1994). ^[5] J. Maddox, *Nature* **335**, 201-201 (1988).

^[6] K. N. Houk, F. Liu, *Acc. Chem. Res.*, **50**, 539 (2017).

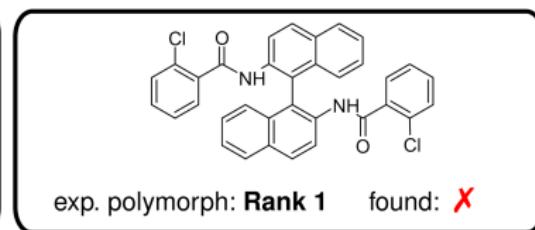
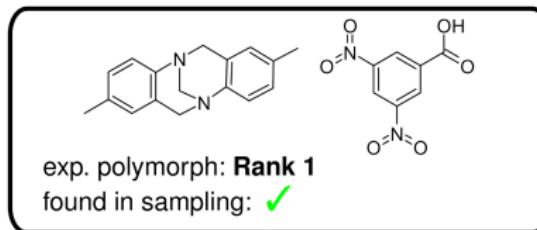
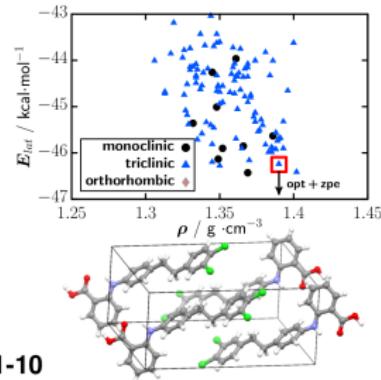
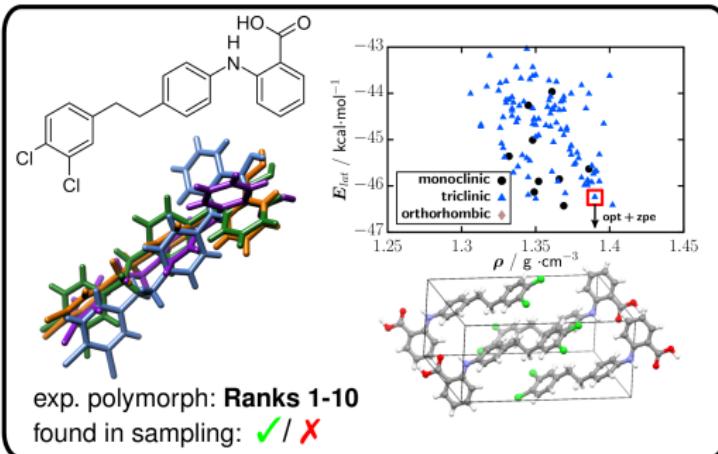
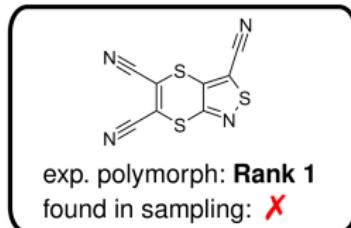
Sampling and energetic ranking for crystal structure prediction



[7] S. Price, *Chem. Soc. Rev.* **43**, 2098 (2014)

[8] M. Vasileiadis, A. V. Kazantsev, P. G. Karamertzanis, C. S. Adjiman, C. C. Pantelides, *Acta Cryst. B* **68**, 677 (2012)

Promising results in the 6th blind test



[9] A. Reilly, et al. *Acta Cryst. B*, **72**, 439 (2016). [10] JGB, S. Grimme, *Acta Cryst. B*, **72**, 502 (2016).

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Metal-free catalysis for H₂ activation

Hydrogen activation



fertilizer



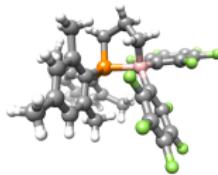
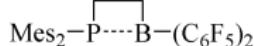
fuel cell



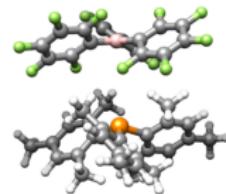
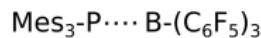
oil cracking

Frustrated Lewis acid and base Pair (FLP)

intra-molecular



inter-molecular



- 'frustration' by carbon bridges (intra) or steric hindrance
- activate small molecules **in solution**^[11–13]

^[11] G. Welch, R. Juan, J. Masuda, D. Stephan, *Science*, **314**, 1124 (2006).

^[12] D. W. Stephan, G. Erker, *Angew. Chem. Int. Ed.* **49**, 46 (2010).

^[13] S. Grimme, H. Kruse, L. Goerigk, G. Erker, *Angew. Chem. Int. Ed.* **49**, 1402 (2010).

FLP condensed phase thermochemistry

		ΔG^{gas}	ΔG^{solv}	ΔG^{solid}
$\text{Mes}_2-\overset{\square}{\text{P}}\cdots\text{B}-(\text{C}_6\text{F}_5)_2$	$+ \text{H}_2 \longrightarrow \text{Mes}_2-\overset{\square}{\underset{\text{H}}{\text{P}}} \text{B}-(\text{C}_6\text{F}_5)_2$	3.9	-5.0	-1.1
$\text{Mes}_2-\overset{\square}{\text{P}}\cdots\text{B}-(\text{C}_6\text{F}_5)_2$	$+ \text{H}_2 \longrightarrow \text{Mes}_2-\overset{\square}{\underset{\text{H}}{\text{P}}} \text{B}-(\text{C}_6\text{F}_5)_2$	10.4	1.6	9.7
$\text{Mes}_2-\overset{\square}{\text{P}}\cdots\text{B}-(\text{C}_6\text{F}_5)_2$	$+ \text{H}_2 \longrightarrow \text{Mes}_2-\overset{\square}{\underset{\text{H}}{\text{P}}} \text{B}-(\text{C}_6\text{F}_5)_2$	11.1	0.5	6.3
$\text{Cy}_3\text{P}\cdots\text{B}(\text{C}_6\text{F}_5)_3$	$+ \text{H}_2 \longrightarrow \text{Cy}_3\text{P} \text{ B}(\text{C}_6\text{F}_5)_3$	-7.2	-18.2	-17.0

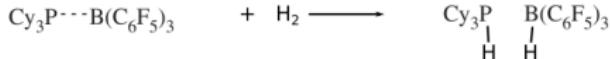
SCAN-D3//HF-3c, HF-3c RRHO energies [kcal/mol]^[14]

COSMO-RS(toluene) implicit solvation, normal conditions

- reaction reversible, H_2 and other small molecules (CO_2 , SO_2)
- two promising candidates for the solid state

^[14] L. Liu, JGB, S. Grimme, *Phil. Trans. R. Soc. A*, **375**, 20170006 (2017).

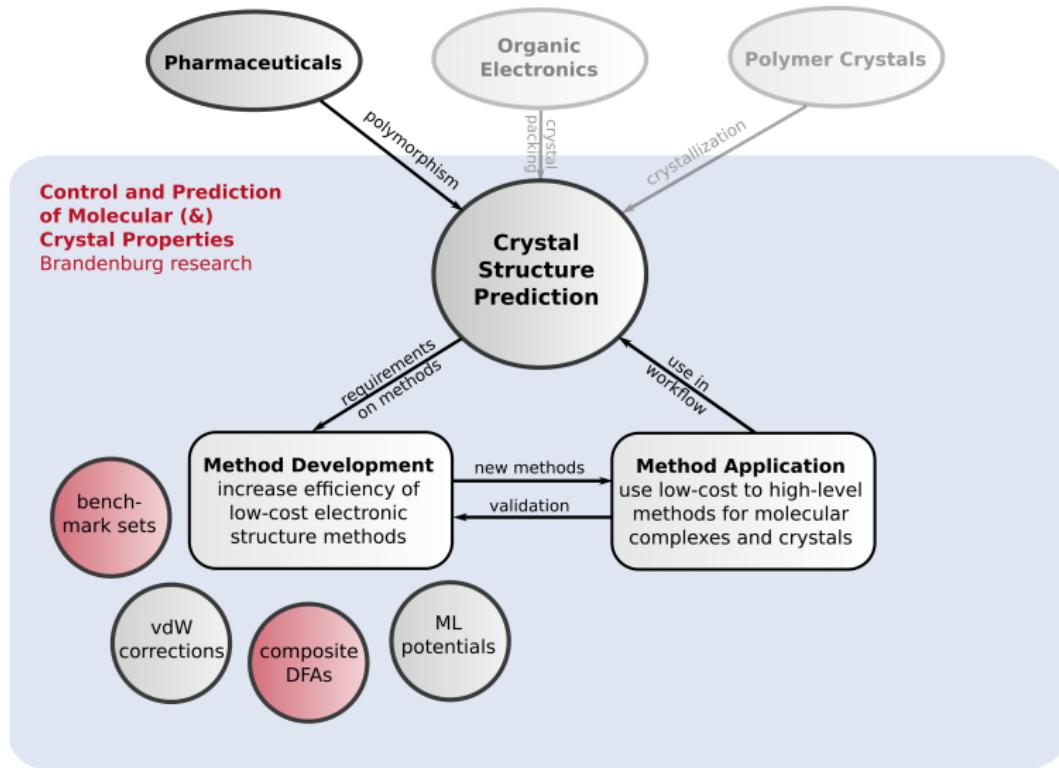
FLP chemistry can be transferred to the solid state



- solid state reactivities confirmed experimentally^[15]
- crystal field stabilizes FLP products
- avoiding undesired products in solution, new FLP routes may be enabled
 - screen for further candidates
 - heterogeneous FLP-catalysis? (work in progress)

^[15] L. Wang, G. Kehr, C. G. Daniliuc, M. Siedow, T. Wiegand, Anna-Lena Wübker, H. Eckert, L. Liu, JGB, S. Grimme, G. Erker submitted (2018).

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Multilevel methodologies: Finding the right compromise

	task/property	example method
accurate QM	single-point energy	many-body WFT (DMC ^[16])
cheap QM	optimization	DFT ^[17–19] (SCAN-D3 ^[20])
very cheap QM	optimization/Hessians conformations	semi-empirical ^[21–22]
force field	dynamics conformational sampling	transferable or molecule specific (QM derived) FF

[16] A. Zen, [JGB](#), J. Klimeš, A. Tkatchenko, D. Alfè, and A. Michaelides, *Proc. Natl. Acad. Sci. U.S.A.*, **115**, 1724 (2018).

[17] [JGB](#), E. Caldeweyher, S. Grimme, *Phys. Chem. Chem. Phys.*, **18**, 15519 (2016).

[18] [JGB](#), C. Bannwarth, A. Hansen, S. Grimme, *JCP*, **148**, 064104 (2018). [19] M. Cutini, et al., *JCTC*, **12**, 3340 (2016).

[20] [JGB](#), J. E. Bates, J. Sun, J. P. Perdew *Phys. Rev. B*, **94**, 115144 (2016).

[21] [JGB](#), S. Grimme, *JPCL*, **5**, 1785 (2014). [22] M. Mortazavi, [JGB](#), R. J. Maurer, A. Tkatchenko, *JPCL*, **9**, 399 (2018)

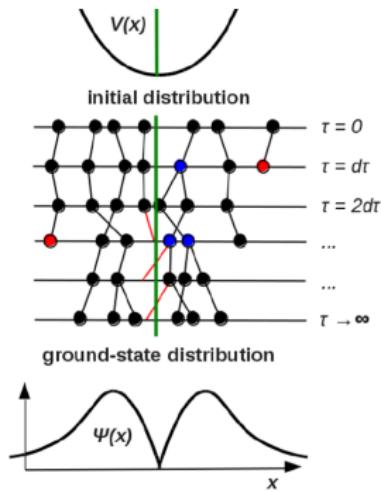
Quantum Monte-Carlo in a nutshell: A scalable high-level method

How to solve Schrödinger equation for large systems?

FN-diffusion Monte-Carlo

$$|\Psi_0(\mathbf{R})\rangle = \lim_{\tau \rightarrow \infty} \exp[-\tau(\hat{H} - E_T)] |\Psi_T(\mathbf{R}, \tau)\rangle$$

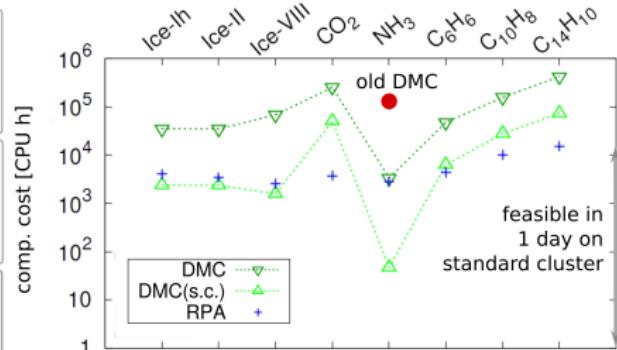
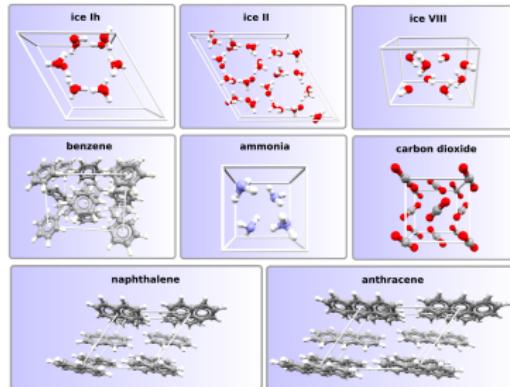
- low-scaling (N^3) with system size
- scaleable to HPCs^[23]
- new size-consistent algorithm leads to substantial speed up^[24]



[23] M. Ďubecký, L. Mitas, P. Jurečka, *Chem. Rev.* **116**, 5188 (2016).

[24] A. Zen, S. Sorella, M. J. Gillan, A. Michaelides, D. Alfé, *Phys. Rev. B* **93**, 241118(R) (2016).

QMC delivers (sub-) chemical accuracy for molecular crystals



- up to three orders of magnitude speed-up compared to best DMC practice two years ago^[25]
- significance will extend to all classes of systems^[16]

[25] L. M. Fraser, W. M. C. Foulkes, G. Rajagopal, R. J. Needs, S. D. Kenny, A. J. Williamson, *Phys. Rev. B* **53**, 1814 (1996).

[16] A. Zen, *JGB*, J. Klimeš, A. Tkatchenko, D. Alfè, A. Michaelides, *Proc. Natl. Acad. Sci. U.S.A.*, **115**, 1724 (2018).

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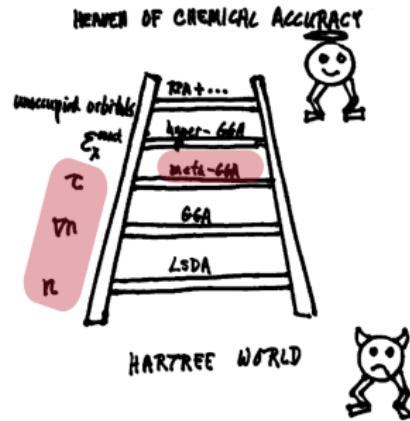
[20] [JGB](#), J. E. Bates, J. Sun, J. P. Perdew *Phys. Rev. B*, **94**, 115144 (2016).

[21] [JGB](#), S. Grimme, *JPCL*, **5**, 1785 (2014). [22] M. Mortazavi, [JGB](#), R. J. Maurer, A. Tkatchenko, *JPCL*, **9**, 399 (2018)

Strongly constrained and appropriately normed semilocal functional (SCAN)

SCAN functional^[26]

- metaGGA expansion
dimensionless $\nabla n, \tau$
- 7 parameters in F_x and F_c
- obeys all 17 known exact constraints
- add semi-classical correction for long-range correlation^[5]

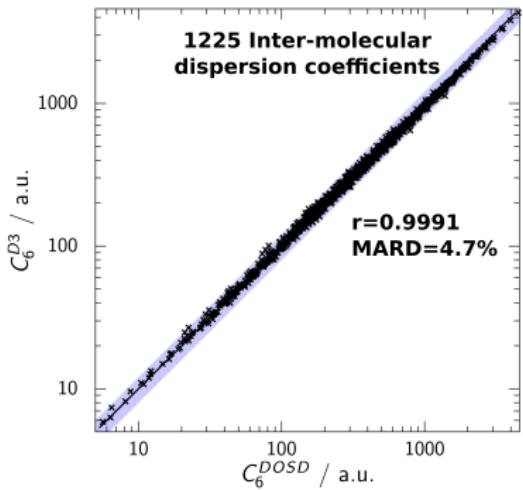


$$E_c^{(D3)} = -\frac{1}{2} \sum_{n=6,8} \sum_{A,B}^{\text{pairs}} \frac{C_n^{AB}}{r_{AB}^n} f_n^d(r_{AB}) - \frac{1}{6} \sum_{A,B,C}^{\text{triples}} \frac{C_9^{ABC} (1 + 3 \cos \theta_A \cos \theta_B \cos \theta_C)}{r_{ABC}^9} f_9^d(r_{ABC})$$

[26] J. Sun, A. Ruzsinszky, J. P. Perdew, *Phys. Rev. Lett.* **115**, 036402 (2015).

[17] JGB, J. E. Bates, J. Sun, J. P. Perdew *Phys. Rev. B*, **94**, 115144 (2016).

Semi-classical correction yields highly accurate dispersion coefficients



D3 correction^[27-29]

$$C_6^{\alpha\beta} = -\frac{3}{\pi} \int_0^{\infty} \alpha^{\alpha}(i\omega) \alpha^{\beta}(i\omega) d\omega$$

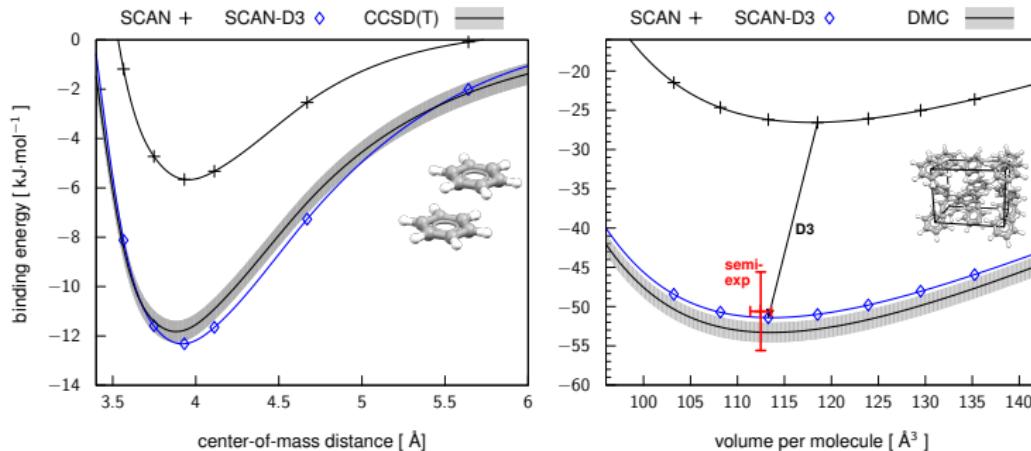
- Casimir-Polder integration of TD-DFT excitations on model hydrides
- residual long-range error < 5%
- empiricism in short-range damping

[27] S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.* **132**, 154104 (2010)

[28] S. Grimme, *WIREs Comput. Mol. Sci.* **1**, 211 (2011)

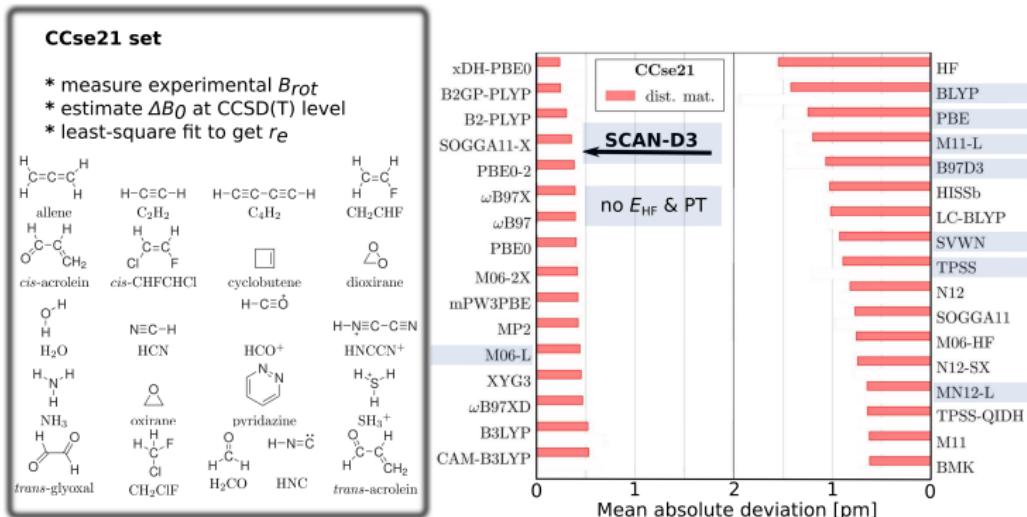
[29] S. Grimme, A. Hansen, JGB, C. Bannwarth, *Chem. Rev.* **116**, 5105 (2016)

Close agreement with reference for noncovalently bound systems



- zero-point and thermal effects crucial for comparing to measurement
- new references valuable for testing approximate methods
- strong hydrogen bonded systems bound to strongly by SCAN

Excellent geometries across the periodic table

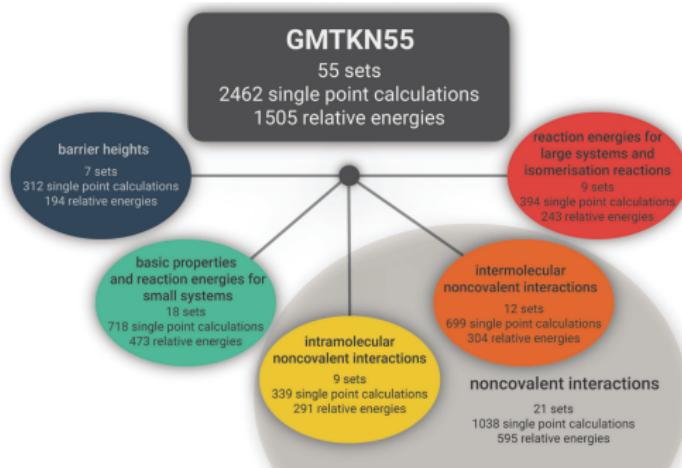


- structures better than all other tested (meta)GGAs
- extends to heavy main group bonds and transition metal complexes

[30] M. Piccardo, E. Penocchio, C. Puzzarini, M. Biczysko, V. Barone, *J. Phys. Chem. A*, **19**, 2058 (2015).

[31] É. Brémond, M. Savarese, N. Q. Su, A. J. Pérez-Jiménez, X. Xu, J. C. Sancho-García, C. Adamo, *JCTC* **12**, 459 (2016).

Reliable main group thermochemistry, kinetics, and noncovalent interactions



- best metGGA functional for GMTKN55, WTMAD-1 of 4.7 kcal/mol
- good performance extends to metal-organic reactions (MOR set)

[¹⁷] JGB, J. E. Bates, J. Sun, J. P. Perdew, *Phys. Rev. B*, **94**, 115144 (2016).

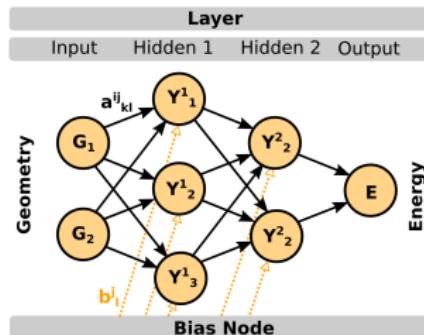
[³²] L. Goerigk, A. Hansen, C. Bauer, S. Ehrlich, A. Najibi, S. Grimme, *Phys. Chem. Chem. Phys.*, **19**, 32184 (2017).

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Understanding dynamical properties of complex systems

- high dimensional neural networks for potential interpolation^[33–34]
- reduce the number of required *ab initio* calculations within CSP
- enable molecular dynamics on approximate *ab initio* potential



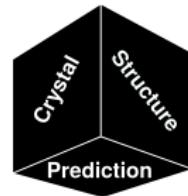
Collaborative Research Centre on
Multiscale Simulation Techniques for Soft Matter

[33] J. Behler, M. Parrinello, *Phys. Rev. Lett.*, **98**, 146401 (2007).

[34] J. Behler, *Angew. Chem. Int. Ed.*, **56**, 12828 (2017).

Building a bridge between molecular and solid state properties

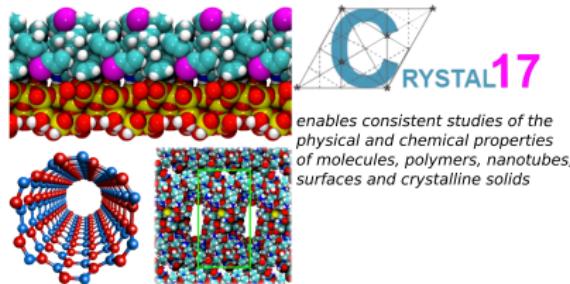
- relative stability of enantiopure and racemic crystal structures
- structure prediction of porous materials
- crystal structure prediction for organic electronics
 - crystal packing effect?
 - modified adsorption/emission?



Extend the collaboration through Merck Lab@TU Darmstadt?

Extend methodologies to inorganic materials

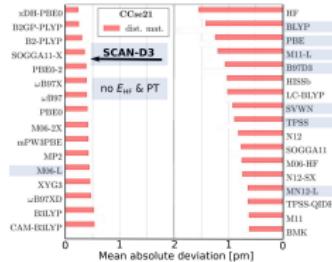
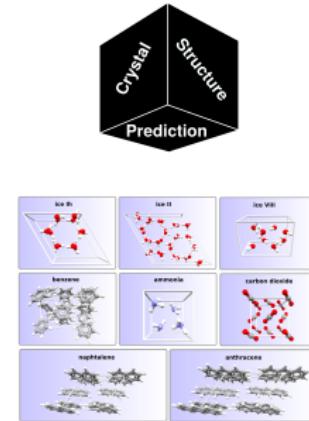
- major contributions to CRYTAL17
- combine efficient program handling with modern density functionals



Organic-inorganic interfaces and nanomaterials?

Summary

- promising results of crystal energy rankings in CSP blind test
- a new principle for FLP chemistry
- DMC delivers (sub-) chemical accuracy with three orders of magnitude speed up
- approximate QM methods like SCAN-D3 useful for reliable electronic structures



Acknowledgements



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Grimme (Bonn)

Principal investigator



Perdew (Temple)



Civalleri (Turin)



Maschio (Turin)

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Michaelides (UCL)



Alf  (UCL)



Tkatchenko
(Luxembourg)



Behler (G ttingen)



Rebecca Sure (BASF)



Luca Iuzzolino



Price (UCL)



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(ISIS)



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Tkatchenko
(Luxembourg)



Behler (G ttingen)



Rebecca Sure (BASF)



Luca Iuzzolino



Price (UCL)



Hall (Bristol)



Fernandez-Alonso
(ISIS)



Erker (M nster)



Antreas Afantitis
(NovaMechanics)

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Stiftung/Foundation

Thanks

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- Solid state thermochemistry

L. Liu, JGB, S. Grimme, *Phil. Trans. A*, **375**, 20170006 (2017).

- QMC for molecular crystals:

A. Zen, JGB, J. Klimeš, A. Tkatchenko, D. Alfè, A. Michaelides, *Proc. Natl. Acad. Sci. U.S.A.*, **115**, 1724 (2018).

- DFT development

JGB, J. E. Bates, J. Sun, J. P. Perdew, *Phys. Rev. B*, **94**, 115144 (2016).

JGB, C. Bannwarth, A. Hansen, S. Grimme *J. Chem. Phys.*, **148**, 64104 (2018).

- website: **gerit-brandenburg.de**