

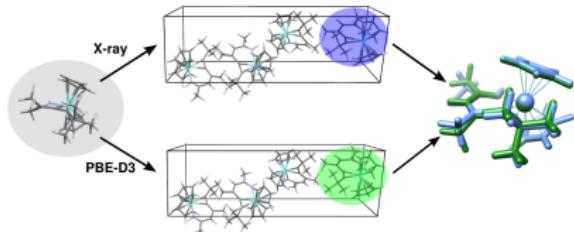
Organic crystal modeling: A hierarchy of quantum chemical methods

Jan Gerit Brandenburg <g.brandenburg@ucl.ac.uk> | 2016-04-15

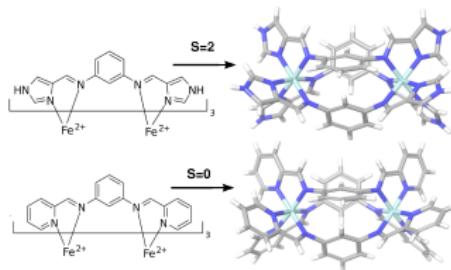
THOMAS YOUNG CENTRE - LUNCHTIME GET TOGETHER - LONDON, UK

Crystal packing dependent properties

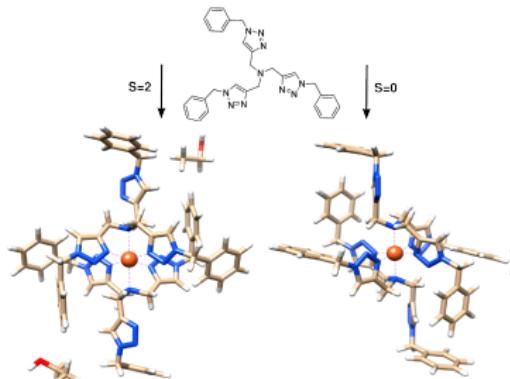
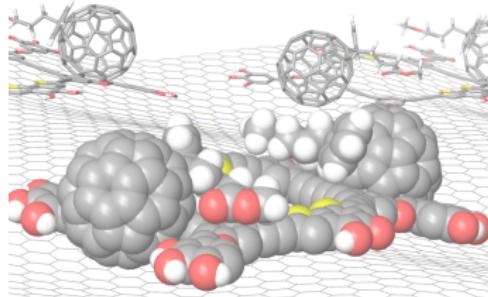
bond isomerization^[1]



spin crossover^[2]



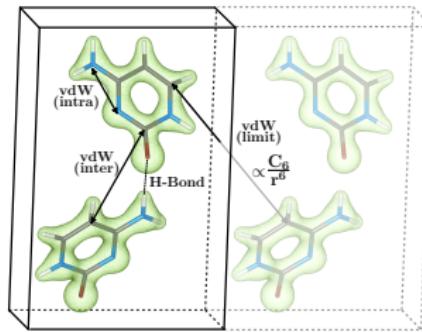
co-adsorption on graphite



^[1]JGB et al., *Organometallics*, 33, 5358 (2014), ^[2]Schweinfurth et al., *Inorg. Chem.*, 53, 8203 (2014)

Intermolecular interactions

- Pauli exchange repulsion
 - electrostatic
 - induction
 - London dispersion
-
- compromise between accuracy and computational efficiency^[3,4,5]



^[3]J. Yang, W. Hu, D. Usvyat, D. Matthews, M. Schütz, G. Chan, *Science*, **345**, 640 (2014)

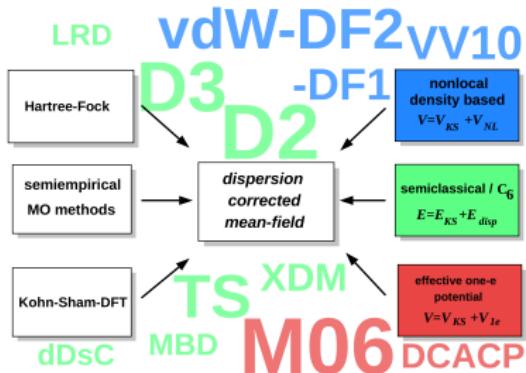
^[4]JGB, S. Grimme, *Top Curr Chem*, **345**, 1 (2014)

^[5]JGB, M. Hochheim, T. Bredow, S. Grimme, *J. Phys. Chem. Lett.* **5**, 4275 (2014)

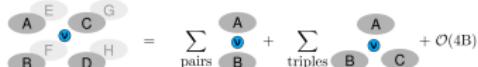
Intermolecular interactions

Formally exact expression vs. practical approaches^[6]

$$E_c = -\frac{1}{2\pi} \int_0^1 d\lambda \int d\mathbf{r} d\mathbf{r}' \frac{1}{|\mathbf{r} - \mathbf{r}'|} \times \int_0^\infty d\omega [\chi_\lambda(\mathbf{r}, \mathbf{r}', i\omega) - \chi_0(\mathbf{r}, \mathbf{r}', i\omega)] ,$$



(1) multi-body expansion



(2) perturbation expansion



(3) multipole expansion

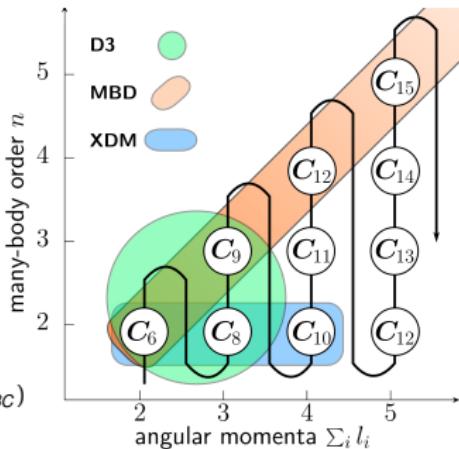


^[6]S. Grimme, A. Hansen, JGB, C. Bannwarth, *Chem. Rev.*, DOI: 10.1021/acs.chemrev.5b00533 (2016)

D3: dispersion correction^[7]

- third order in many-body expansion
- quadrupole terms in multipole expansion

$$E_{\text{disp}}^{(\text{D3})} = -\frac{1}{2} \sum_{n=6,8} \sum_{A,B}^{\text{pairs}} \frac{C_n^{AB}}{r_{AB}^n} \cdot f_n^d(r_{AB}) \\ - \frac{1}{6} \sum_{A,B,C}^{\text{triples}} \frac{C_9^{ABC} (1 + 3 \cos \theta_A \cos \theta_B \cos \theta_C)}{r_{ABC}^9} \cdot f_9^d(r_{ABC})$$



gCP: geometric counterpoise correction^[8,9]

^[7] S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.*, **132**, 154104 (2010)

^[8] H. Kruse, S. Grimme, *J. Chem. Phys.*, **136**, 154101 (2012)

^[9] JGB, M. Alessio, B. Civalleri, M. Peintinger, T. Bredow, S. Grimme, *JPC A*, **117**, 9282 (2013)

Multilevel methodologies

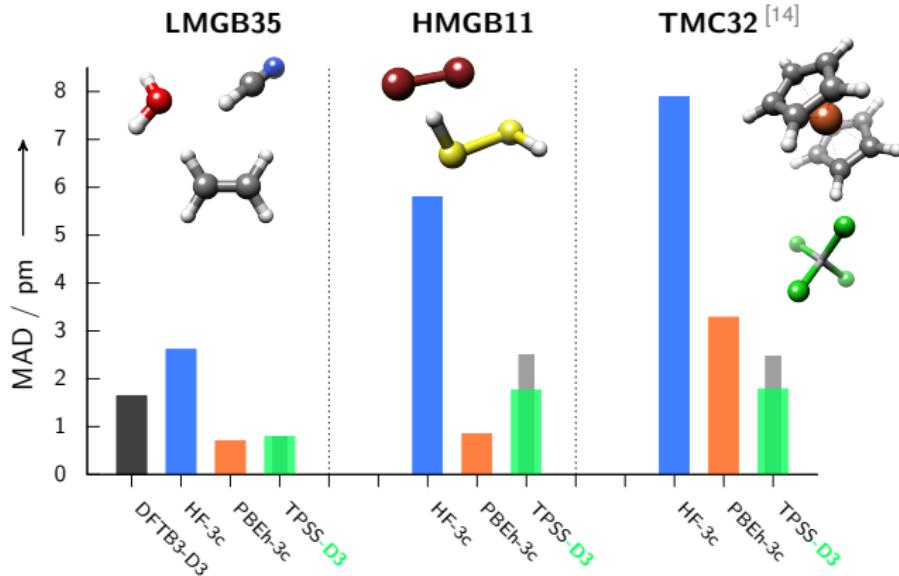
	basis set	DISP	BSSE	additional
TPSS-D3	'CBS'	D3 ^{atm}	–	–
	$\downarrow \times 10^1$ speed-up			
PBEh-3c^[10] & HSE-3c^[11]	DZ	D3 ^{atm}	gCP	mod. E_{xc}
	$\downarrow \times 10^1$			
HF-3c^[12]	minimal	D3 ^{atm}	gCP	SRB
	$\downarrow \times 10^2$			
DFTB3-D3^[13]	minimal	D3 ^{atm}	–	HX-damping, SK splines

^[10] S. Grimme, JGB, C. Bannwarth, A. Hansen, *J. Chem. Phys.*, **143**, 054107 (2015)

^[11] JGB, E. Caldeweyher, S. Grimme, *PCCP*, (submitted) ^[12] R. Sure, S. Grimme, *J. Comput. Chem.*, **34**, 1672 (2013)

^[13] JGB, S. Grimme, *J. Phys. Chem. Lett.* **5**, 1785 (2014)

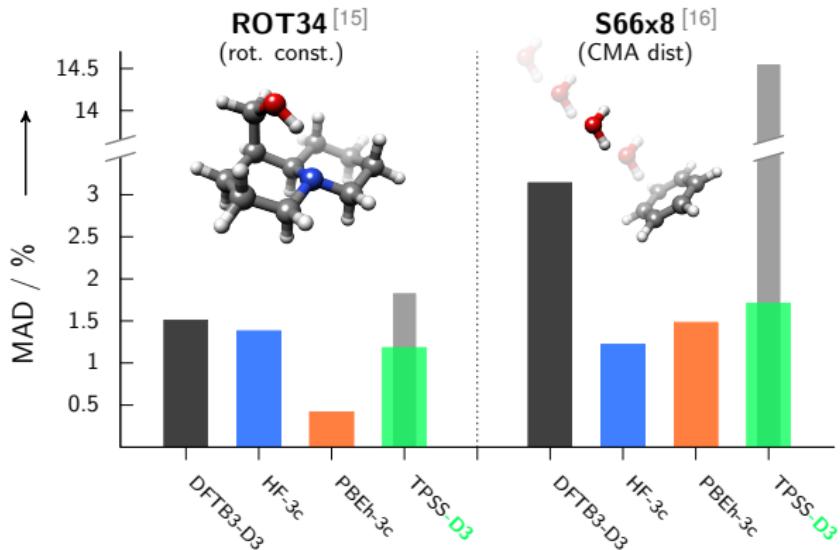
Covalent bond lengths



- Fock exchange problematic for 3d complexes

[14] M. Bühl, H. Kabrede, *J. Chem. Theory Comput.*, **2**, 1282 (2006)

Noncovalent bonds length



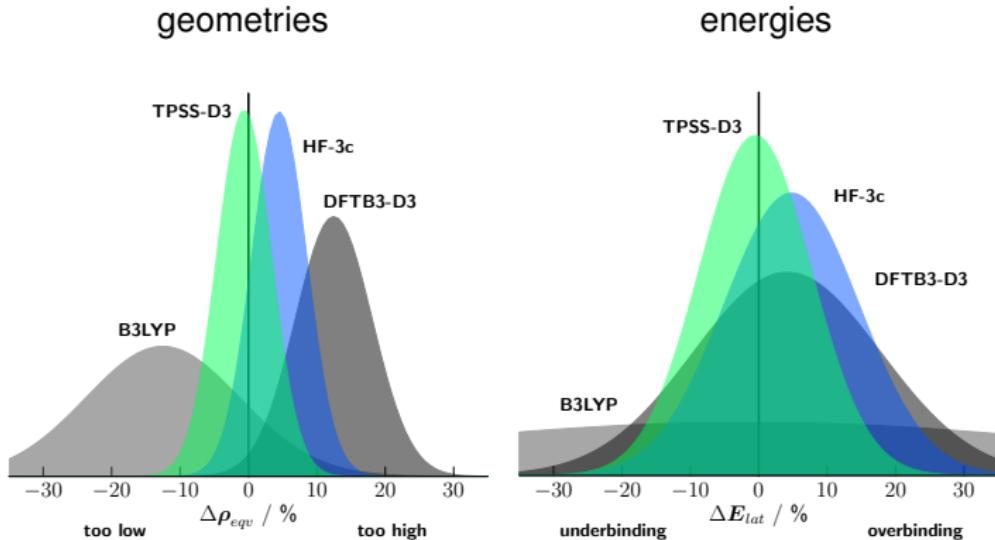
- significant effect of dispersion interaction

[¹⁵] S. Grimme, M. Steinmetz, *Phys. Chem. Chem. Phys.*, **15**, 16031 (2013)

[¹⁶] J. Rezáč, K. Riley, P. Hobza, *J. Chem. Theory Comput.*, **8**, 2427 (2011)

NCI: molecular crystals

X23^[17,18] and ICE10^[19] solid state benchmark sets

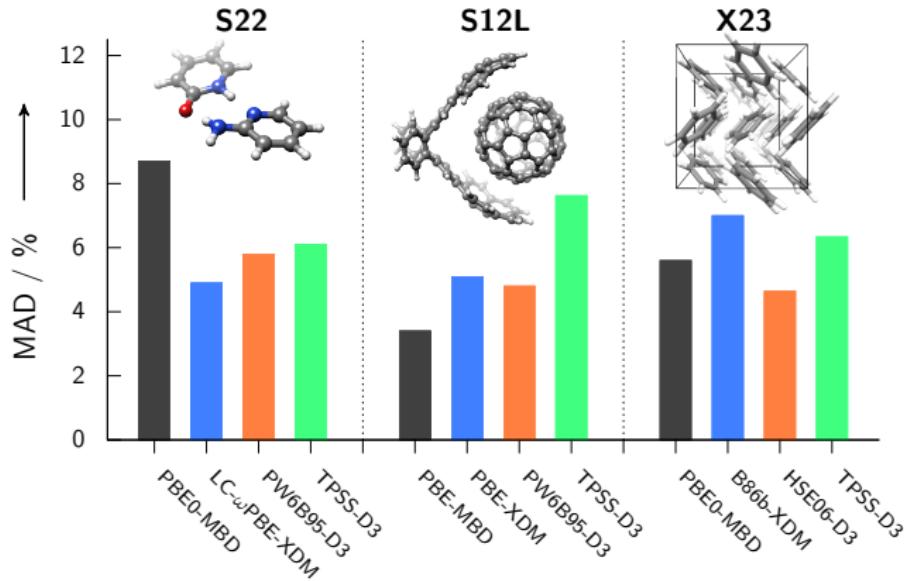


[¹⁷] E. Johnson, A. Otero-de-la-Roza, *J. Chem. Phys.* **137**, 054103 (2012)

[¹⁸] A. Reilly, A. Tkatchenko, *J. Chem. Phys.* **139**, 024705 (2013)

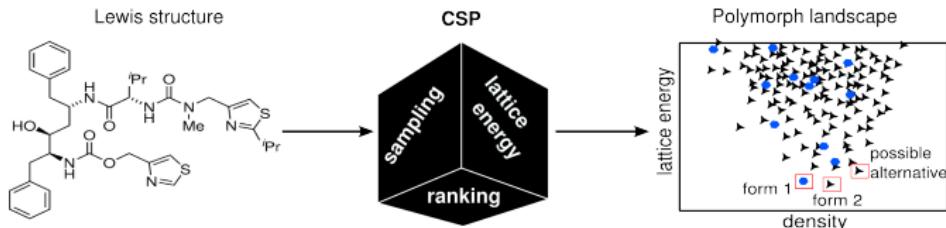
[¹⁹] JGB, T. Maas, S. Grimme, *J. Chem. Phys.* **142**, 124104 (2015)

Performance of dispersion models



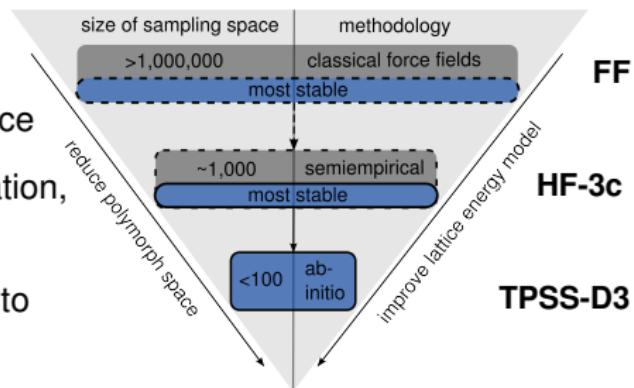
- error in S12L and X23 reference about 3-5%
- different dispersion corrected DFAs yield high quality results

Crystal structure prediction challenge



Layers of complexity:

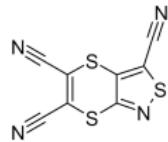
- molecular conformational space
- space groups (varying orientation, conformation, etc.)
- polymorph ranking according to free energy^[20,21]



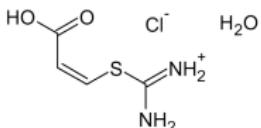
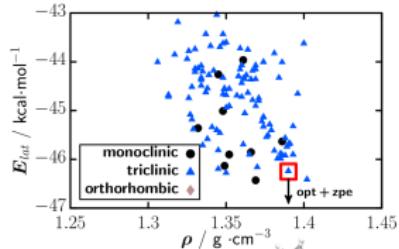
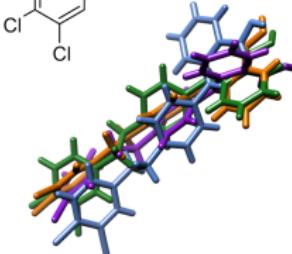
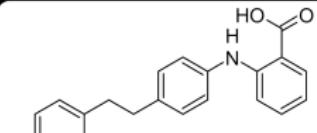
[20] S. Price, *Chem. Soc. Rev.* **43**, 2098 (2014)

[21] M. Neumann, F. Leusen, J. Kendrick, *Angew. Chem. Int. Ed.* **47**, 2427 (2008)

Blind test results

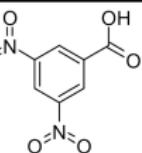
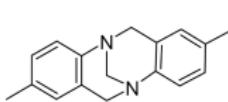


exp. polymorph: **Rank 1**
found in sampling: **X**

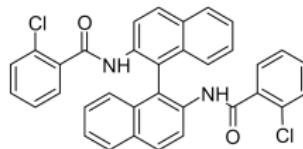


exp. polymorph: **Rank 1**
found in sampling: **X**

exp. polymorph: **Ranks 1-10**
found in sampling: **✓ / X**



exp. polymorph: **Rank 1**
found in sampling: **✓**



exp. polymorph: **Rank 1** found: **X**

[22] A. Reilly, et al. *Acta Cryst. B*, submitted

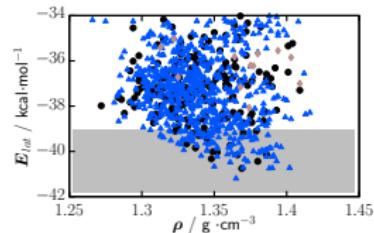
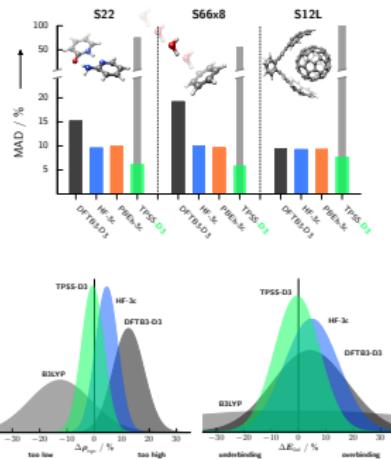
Summary

Conclusions

- hierarchy of electronic structure methods
- simplified methods could bridge the gap between FF and DFT
- HF-3c and TPSS-D3 for crystal structure prediction

Outlook and possible improvements

- apply higher levels for final ranking
- improve FF → HF-3c transition



Acknowledgements



Collaborators

- **Stefan Grimme** (Bonn)
- Thomas Bredow (Bonn)
- Gerhard Erker (Münster)
- Bartolomeo Civalleri (Torino)
- Sally Price (London)
- Anthony Reilly (Cambridge)

Funding



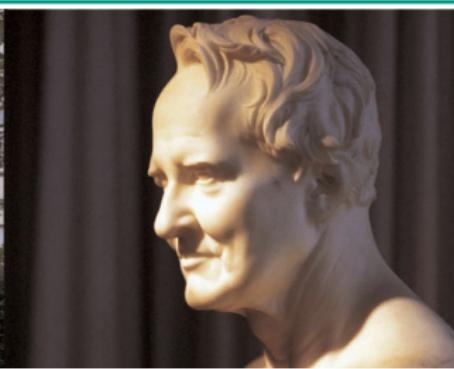
Chemie an Spinzentren
SFB 813 der Deutschen
Forschungsgemeinschaft





AvH goals

- connecting academic excellence worldwide
- knowledge transfer and cooperation at the highest level
- strengthening cutting-edge research through internationalization



Alexander von Humboldt foundation



Alexander von Humboldt
Stiftung/Foundation

postdoctoral researchers

PhD completed less than 4 years ago

- Research fellowship: 6-24 month with German host

junior group leaders

PhD completed less than 6 years ago

- Sofia Kowalewskaja award: own research group in Germany



Alexander von Humboldt foundation



Alexander von Humboldt
Stiftung/Foundation

experienced researcher

PhD completed less than 12 (18) years ago

- Research fellowship: 6-18 month with German host
- Friedrich Wilhelm Bessel Award: research stay in Germany

international cutting-edge researcher

- Humboldt research award and Alexander von Humboldt Professorship





Contact

AvH in Bonn, Germany

■ info@avh.de

■ www.humboldt-foundation.de

AvH Association, UK

■ info@avh.org.uk

■ www.avh.org.uk

JGB: g.brandenburg@ucl.ac.uk www.gerit-brandenburg.de

