

Predicting properties of molecular crystals by multilevel strategies

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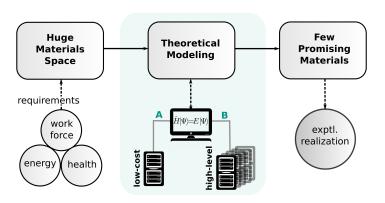
Outline of talk



- Introduction
- Quantum Monte-Carlo for molecular materials
- 3 HSE-3c: A low-cost electronic structure method
- Simulation based crystal structure prediction
- 5 Conclusions

Materials discovery can employ computational models





A: approximate models & local computer cluster

B: high-level models & world leading computational facility

Exact simulation of extended systems computationally very demanding

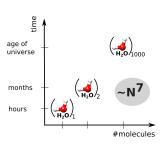


Paul Adrian Maurice Dirac (1902-1984)



"The underlying physical laws (...) of a large part of physics and the **whole of chemistry** are thus completely known,

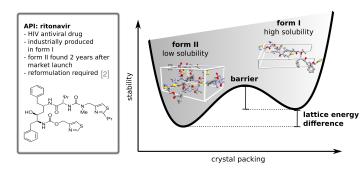
and the difficulty is only that the exact application of these laws leads to equations much too complicated to be soluble"[1]



^[1]P. A. M. Dirac, *Proc. Roy. Soc.Ser. A* **123**, 714 (1929)

Academic and industrial interest in molecular materials





- tools to predict possible polymorphs would be valuable^[3]
 - → Currently no high-level method applicable

[3] S. L. Price, JGB, Molecular Crystal Structure Prediction; Elsevier Australia ISBN: 9780128098356 (2017).

^[2] J. Bauer, et al., J. Pharm. Res. 18, 859-866 (2001).

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Quantum Monte-Carlo in a nutshell:



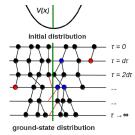
A scalable high-level method

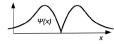
Fixed-node diffusion Monte-Carlo

- Enforce nodal surface of Fermions $\Gamma = \{\mathbf{R} : |\Psi_{\tau}\rangle = 0\}$
- Walkers in configuration space $|\Psi_T(\mathbf{R}, au)
 angle = \operatorname{hist}\left[\sum \delta(\mathbf{R}-\mathbf{R_i}(au))
 ight]$
- 3) Diffusion in imaginary time

$$\partial_{ au} |\Psi_{T}(\mathbf{R}, au)
angle = \left[\frac{1}{2} \nabla_{\mathbf{R}}^{2} - (V - E_{T}) \right] |\Psi_{T}(\mathbf{R}, au)
angle$$

4) Projection to exact ground state $|\Psi_0(\mathbf{R})\rangle = \lim_{\epsilon \to 0} \exp\left[-\tau(\hat{H} - E_T)\right] |\Psi_T(\mathbf{R}, \tau)\rangle$



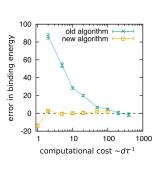


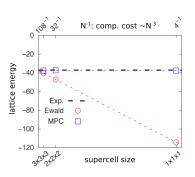
- low-scaling (N^3) with system size
- scalable to high-performance computing facilities

[4] M. Ďubecký, L. Mitas, P. Jurečkaâ, Chem. Rev. 116, 5188 (2016).

New QMC algorithm leads to substantial speed up







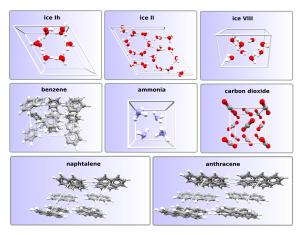
- lacktriangle new size-consistent implementation reduces Δau error drastically $^{ extstyle [5]}$
- Model periodic Coulomb for finite size correction [6]

^[5] A. Zen, S. Sorella, M. J. Gillan, A. Michaelides, D. Alfé, *Phys. Rev. B* 93, 241118(R) (2016).

^[6] L. M. Fraser, W. M. C. Foulkes, G. Rajagopal, R. J. Needs, S. D. Kenny, A. J. Williamson, Phys. Rev. B 53, 1814 (1996).

Diverse interactions in test cases





- strong H-bonds, vdW of saturated and unsaturated molecules
- problematic for all readily applicable methods (DFT-D, MP2)

8/23

QMC delivers (sub-) chemical accuracy for all tested systems



{non-published data}

- excellent agreement with experiment and CCSD(T)^[7]
- uncertainty in H_{sub}^{exp} probably larger than DMC errors

[7] Y. S. Al-Hamdani, M. Rossi, D. Alfè, T. Tsatsoulis, B. Ramberger, <u>JGB</u>, A. Zen, G. Kresse, A. Grüneis, A. Tkatchenko, A. Michaelides *J. Chem. Phys.* **147**, 044710 (2017).

QMC is feasible within one day on standard computer cluster



{non-published data}

up to three orders of magnitude speed-up compared to best DMC practice two years ago[8]

[8] A. Zen, JGB, J. Klimeš, A. Tkatchenko, D. Alfè, A. Michaelides, submitted

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Lessons learned in the past years



- realistic structures are key to many important physical and chemical properties
- Kohn-Sham density functional theory is method of choice for structures;
 wavefunction methods may take over for energies
- configurational sampling, entropy, and solvation issues are as important as good convergence in electronic energy

Multilevel methodologies: Finding the right compromise



	task/property	example method
accurate QM	single-point energy	DLPNO-CCSD(T) DMC, RPA+SE
cheap QM	optimization	metaGGA (SCAN-D3 ^[9]) HSE-3c ^[10,11]
very cheap QM	optimization/Hessians conformations	semi-empirical HF-3c $^{[12,13]}$, DFTB3-D3 $^{[14]}$
force field	dynamics conformational sampling	transferable or molecule specific (QM derived) FF

^{[9] &}lt;u>JGB</u>, J. E. Bates, J. Sun, J. P. Perdew *Phys. Rev. B*, **94**, 115144 (2016)

^{[10] &}lt;u>JGB,</u> E. Caldeweyher, S. Grimme, *Phys. Chem. Chem. Phys.*, **18**, 15519 (2016)

^[11] S. Grimme, <u>JGB</u>, C. Bannwarth, A. Hansen, *J. Chem. Phys.*, **143**, 054107 (2015)

^[12] R. Sure, S. Grimme, J. Comput. Chem., 34, 1672 (2013) [13] <u>JGB</u>, S. Grimme, Top. Curr. Chem, 345, 1 (2014)

^[14] JGB, S. Grimme, J. Phys. Chem. Lett. 5, 1785 (2014)

HSE-3c: Small basis DFT with semi-classical potentials



Requirements

- \sim 10 \times faster vs. standard DFA → small atomic orbital expansion
- reduce self-interaction error → use Fock exchange
- numerically robust → long-range screening of exchange

$$E_{\mathrm{tot}}^{\mathrm{HSE-3c}} = E^{\mathrm{(modHSE)}} + E_{\mathrm{DISP}}^{\mathrm{(D3)}} + E_{\mathrm{BSSE}}^{\mathrm{gCP}}$$

Technical details:[10]

- modified HSE^[15] in small def2-mSVP^[11] basis set
- D3 and gCP semi-classical corrections (7 global parameters)

[10] JGB, E. Caldeweyher, S. Grimme, Phys. Chem. Chem. Phys., 18, 15519 (2016)

[11] S. Grimme, JGB, C. Bannwarth, A. Hansen, J. Chem. Phys., 143, 054107 (2015)

[15] J. Heyd, G. E. Scuseria, M. Ernzerhof, J. Chem. Phys. 124, 219906 (2006)

Compromise of known functionals for exchange correlation functional



$$E_{ ext{xc}}^{ ext{(modHSE)}} = \mathbf{a_x} \, E_{ ext{x}}^{ ext{(HF,SR)}}(\omega) + (1-a_{ ext{x}}) \, E_{ ext{x}}^{ ext{(HSE,SR)}}(\omega) + E_{ ext{x}}^{ ext{(HSE,LR)}}(\omega) + E_{ ext{c}}^{ ext{(modPBE)}}$$

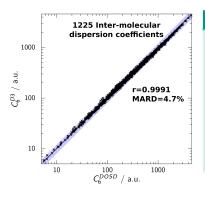
modified HSE to reproduce modified PBE-XC

$$F_X^{\mathrm{PBE}} = 1 + \frac{\mu s}{1 + \frac{\mu s^2}{\kappa}}, \qquad \qquad s = |\nabla \rho / \rho^{4/3}|$$

- lacksquare from PBEsol, κ averaged from PBE/revPBE
- $f \beta = 0.03$ in $F_C^{\rm PBE}$ fitted to atomization energies
- $a_x = 0.42$: getting bond length right (standard range-separation $\omega = 0.11$)
- mSVP atomic orbitals fixed and available for whole PES
 - → only seven globally fitted parameters

Semi-classical correction yields highly accurate dispersion coefficients





D3 correction[16-18]

$$C_6^{\alpha\beta} = -\frac{3}{\pi} \int_0^\infty \alpha^\alpha (i\omega) \alpha^\beta (i\omega) \,\mathrm{d}\omega$$

- Casimir-Polder integration of TD-DFT excitations on model hydrides
- residual long-range error < 5%
- empiricism in short-range damping

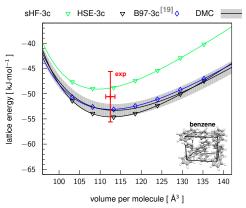
^[16] S. Grimme, J. Antony, S. Ehrlich, H. Krieg, J. Chem. Phys. 132, 154104 (2010)

^[17] S. Grimme, WIREs Comput. Mol. Sci 1, 211 (2011)

^[18] S. Grimme, A. Hansen, JGB, C. Bannwarth, Chem. Rev. 116, 5105 (2016)

Close agreement with reference for equation of state of solid benzene





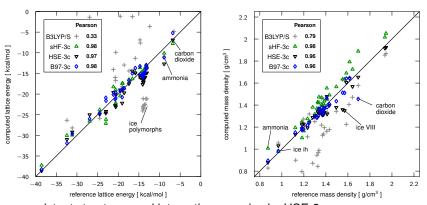
- zero-point and thermal effects crucial for comparing to measurement
- new references valuable for testing approximate methods

[19] JGB, C. Bannwarth, A. Hansen, S. Grimme, submitted.

Good results on molecular crystals



X23^[20,21] and ICE10^[22] benchmark sets:



consistent structures and interaction energies by HSE-3c

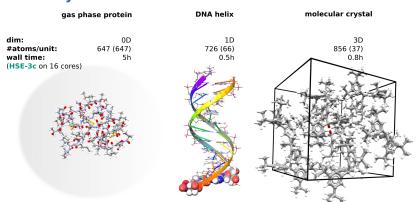
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^[20] E. Johnson, A. Otero-de-la-Roza, *J. Chem. Phys.* 137, 054103 (2012), [21] A. Reilly, A. Tkatchenko, *JCP* 139, 024705 (2013)

^[22] JGB, T. Maas, S. Grimme, J. Chem. Phys. 142, 124104 (2015)

Fast electronic structure with ab-initio accuracy





- fast computer code CRYSTAL17^[23] with cost-efficient methods
- enabling routine electronic structure calculation of large systems

R. Dovesi, et al., Int. J. Quantum Chem., 114, 1287-1317 (2014), new release in 2017

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Predict most stable crystal polymorphs based on the molecular diagram



The inability to predict something as simple as how a molecule would crystallize is one of the continuing scandals in the physical sciences. [24,25]

Task

- molecule is chosen due to its chemical/physical/biological properties
- based on the molecular diagram only, the most stable crystal structures should be predicted
- predict properties of interest for the most promising candidates

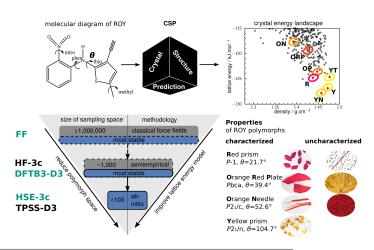
[24] A. Gavezzotti, Acc. Chem. Res. 27, 309-314 (1994).

[25] J. Maddox, Nature 335, 201-201 (1988).

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Sampling and energetic ranking for crystal structure prediction





[26] S. Price, Chem. Soc. Rev. 43, 2098 (2014)

[27] M. Vasileiadis, A. V. Kazantsev, P. G. Karamertzanis, C. S. Adjiman, C. C. Pantelides, Acta Cryst. B 68, 677 (2012)

Introduction QMC for molecular materials

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Promising results in the 6th blind test











	22	23	24	25	26
PBE	2	1-9	6	3	1
PBE-D3	1	1-9	1	1	- 1
PBE-MBD	1	1-7	1	2	1
vdW-DF2	1	4-8	1	3	2
M06L	1	4-13	1	1	7

lattice energy on fixed TPSS-D3 structures

HSF-3c: A low-cost method

- good lattice energy based ranking of PBE-D3^[28,29]
- some structures lost in FF → DFT transition

[28] A. Reilly, et al. Acta Cryst. B, 72, 439 (2016)

[29] JGB, S. Grimme Acta Cryst. B, 72, 502 (2016)

Summary

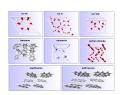


Conclusions

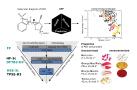
- DMC delivers (sub-) chemical accuracy with three orders of magnitude speed up
- cheap QM methods like HSE-3c useful for fast electronic structures
- promising results of crystal energy rankings in CSP blind test

Outlook

- extending the merits of DFT-D in CSP
- explore systems like APIs, OLEDs, MOFs, where high-level accuracy is needed
 - → ritonavir polymorphs running







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Collaborators

- Andrea Zen (London)
- Jiří Klimeš (Prague)
- Alexander Tkatchenko (Luxenburg)
- Dario Alfè (London)
- Angelos Michaelides (London)

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- Felix Fernandez-Alonso (Harwell Oxford)
- Bartolomeo Civalleri (Torino)
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- Stefan Grimme (Bonn)







Key references



QMC for molecular crystals:

A. Zen, JGB, J. Klimeš, A. Tkatchenko, D. Alfè, A. Michaelides, submitted.

DFT development

<u>JGB</u>, J. E. Bates, J. Sun, J. P. Perdew, *Phys. Rev. B*, **94**, 115144 (2016). <u>JGB</u>, E. Caldeweyher, S. Grimme, *Phys. Chem. Chem. Phys.*, **18**, 15519 (2016).

DFA-DISP for crystal structure prediction:

S. Grimme, A. Hansen, <u>JGB</u>, C. Bannwarth, *Chem. Rev.* **116**, 5105 (2016). S. L. Price, <u>JGB</u>, *Molecular Crystal Structure Prediction*, G. DiLabio, A. Otero-de-la-Roza, Eds., Elsevier Australia, *ISBN*: *9780128098356* (2017).

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