

# Nailing down the water-graphene adsorption strength indicates a mild hydrophilicity

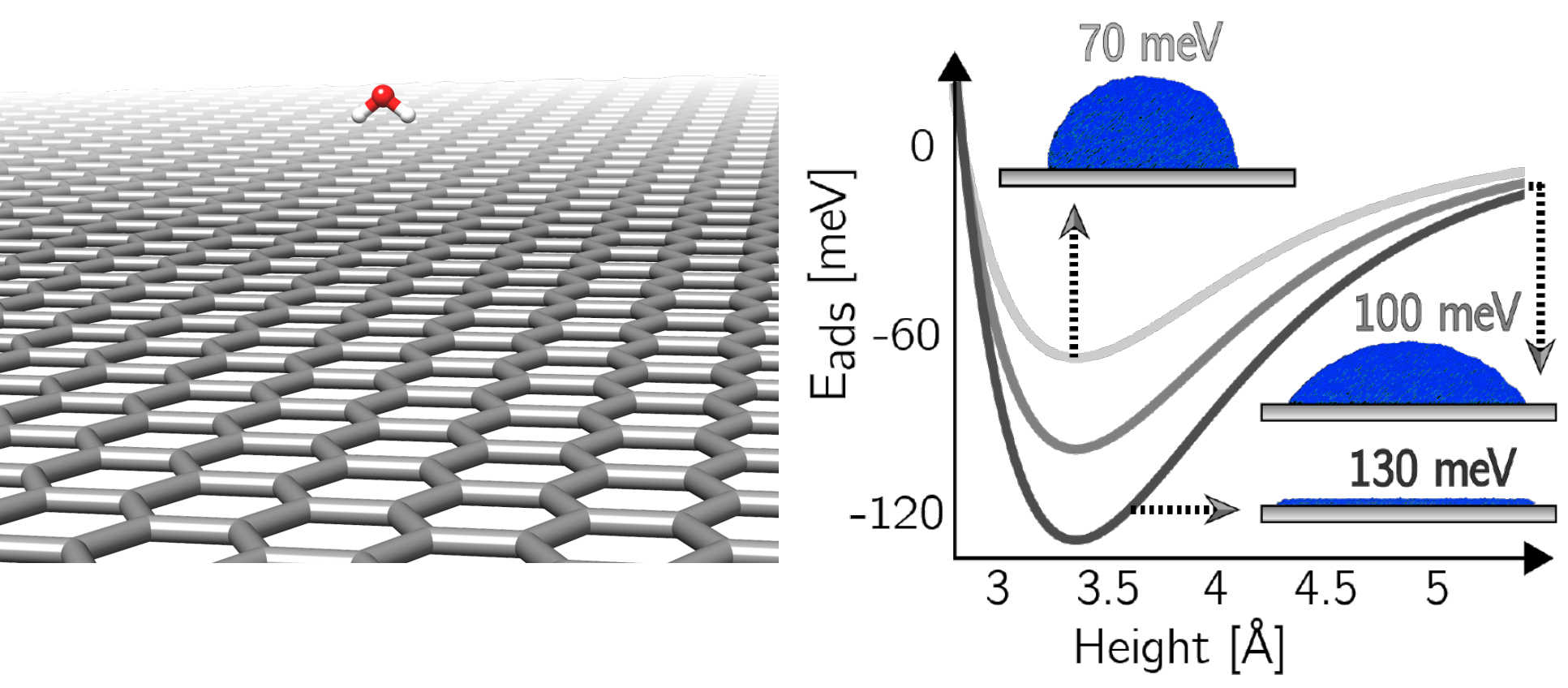
## Physisorption of Water on Graphene: Subchemical Accuracy from Many-Body Electronic Structure Methods

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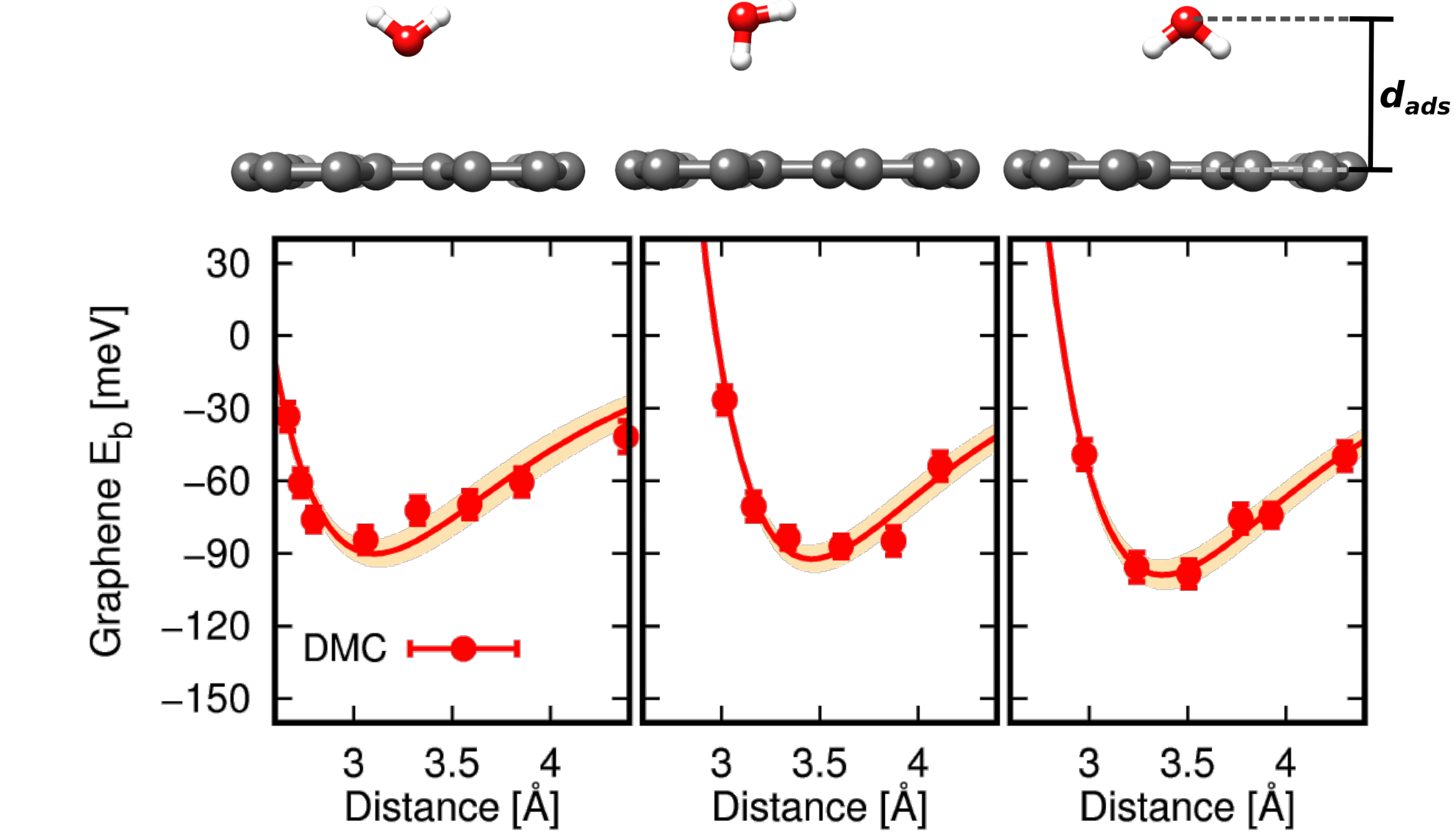
### Background

- Is graphene hydrophobic or hydrophilic?
- Directly related to adsorption.

E <sub>b</sub> [meV]	Method	Comment
-70 ± 10	DMC	Finite-size errors
-135	i-CCSD(T)	Small basis set
-80 ± 100	DFT	Highly XC dependent
-87	p-CCSD(T)	Large basis set
-99 ± 6	DMC	FSE corrected



### RESULTS



- Motif independence → low friction.
  - Contact angle is  $\leq 56^\circ$ .
- Graphene is mildly hydrophilic.<sup>[1]</sup>

### METHODS

- Fixed-node Diffusion Monte-Carlo.<sup>[2]</sup>
- (Periodic) CCSD(T), RPA.
- 5x5 supercell of graphene, bias from finite coverage and size corrected.

### AMMO BAR

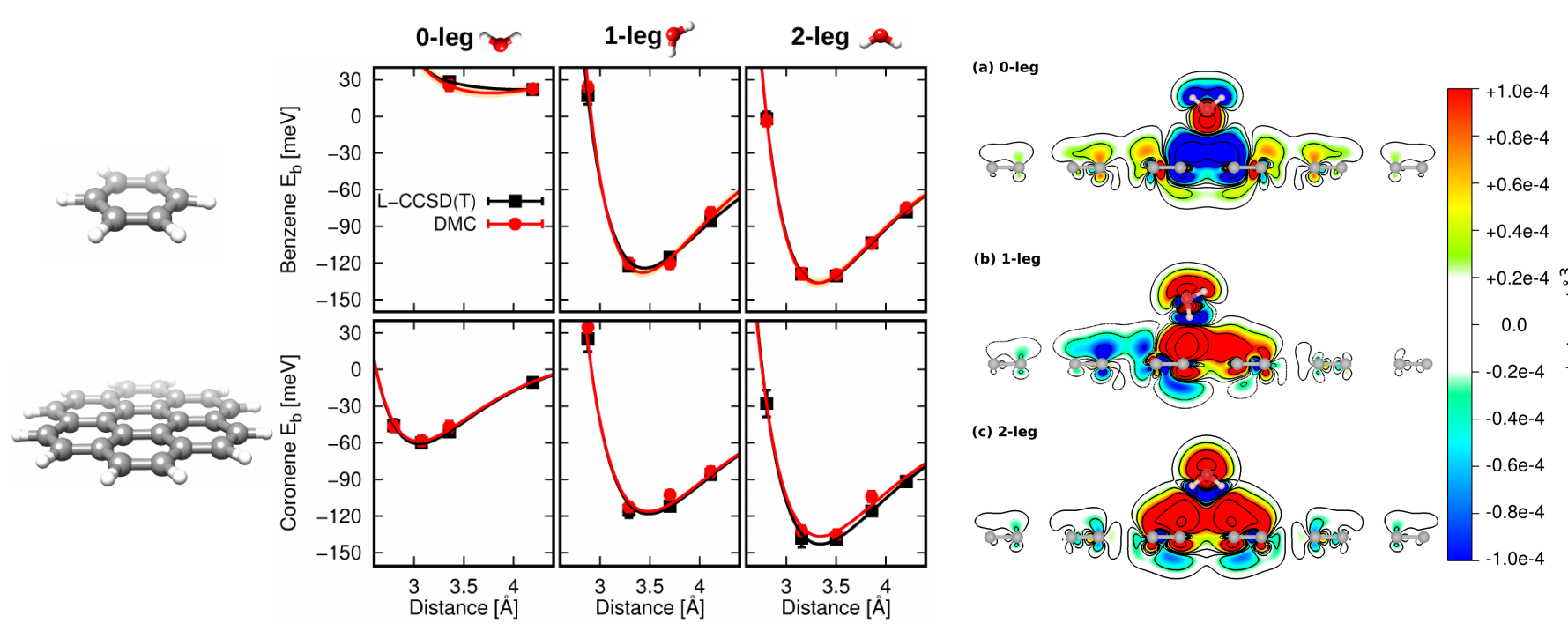


TABLE I. Reference interaction energies for considered systems given per water molecule in meV. The reported error represents the uncertainty on the evaluation.

w@graphene <sup>55</sup>	w@benzene <sup>55</sup>	w@coronene <sup>55</sup>
0-leg -90 ± 6	0-leg -43 ± 1	0-leg -61 ± 3
1-leg -92 ± 6	1-leg -124 ± 3	1-leg -118 ± 5
2-leg -99 ± 6	2-leg -136 ± 2	2-leg -143 ± 4

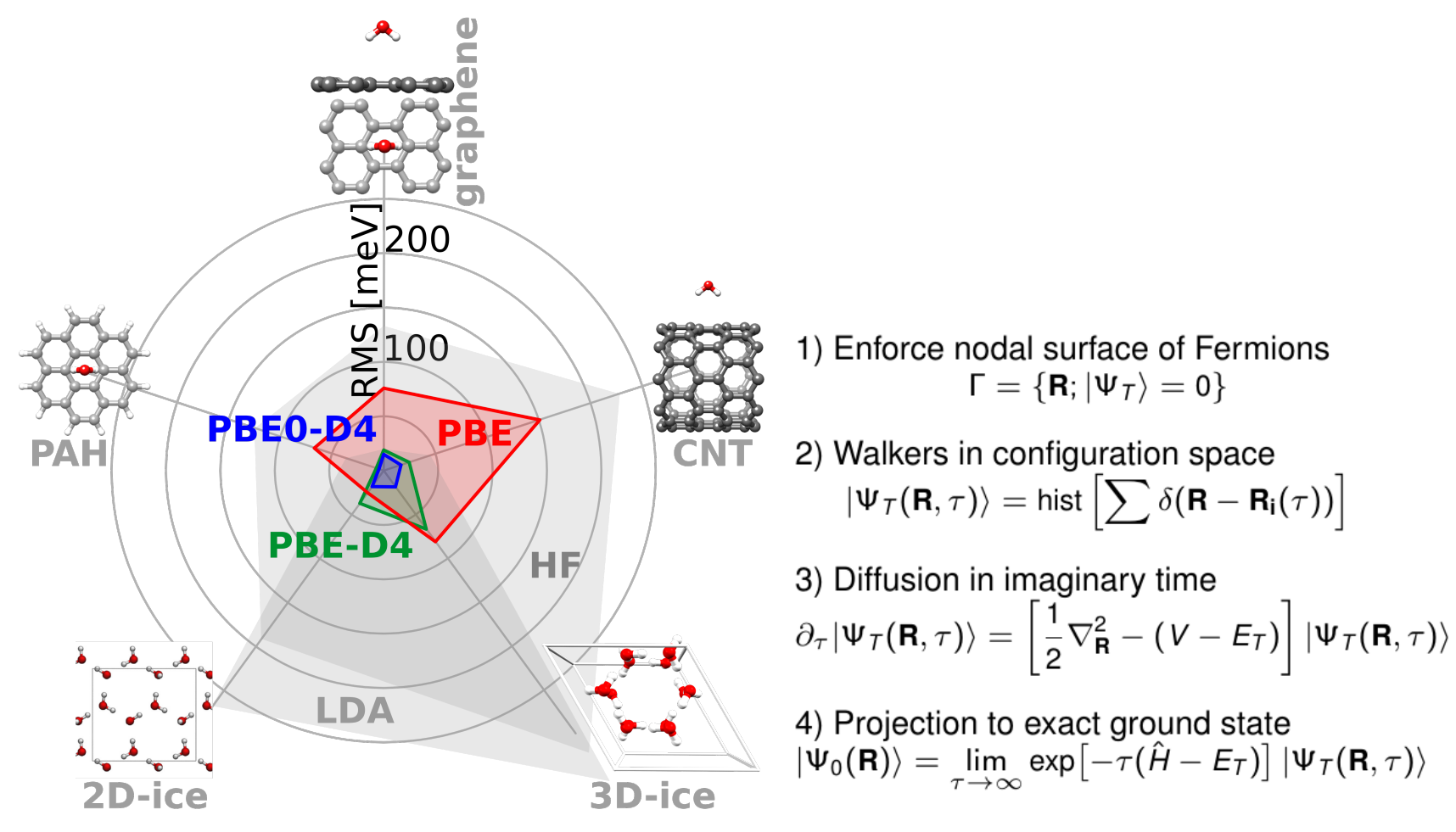
  

w@CNT <sup>54</sup>	external	internal
-85 ± 18	-85 ± 18	-287 ± 16

3D ice <sup>42</sup>	2D ice <sup>17</sup>	pent
Ih -615 ± 5	hex -423 ± 1	-419 ± 3
II -613 ± 6	square -404 ± 3	rhombic -389 ± 3
VIII -594 ± 6		

<sup>a</sup> Evaluation based on Ref. 52, but recomputed in this work. See details in Section II B.

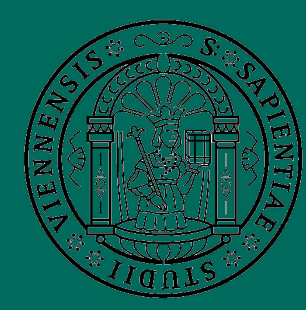


**DMC Details:**  
Fock pseudopotentials with the LA are used. The trial wavefunctions were of the Slater-Jastrow type with LDA determinant. The recently developed size-consistent DMC algorithm (ZSGMA) was used. Finite time-step errors are carefully minimised by performing simulations with different values of the time-step, till the bias appears safely smaller than the stochastic error. In periodic calculations, finite-size corrections are applied using the model periodic Coulomb interaction.

[1] *J. Phys. Chem. Lett.* **2019**, 10, 358  
[2] *P.N.A.S. USA* **2018**, 115, 1724



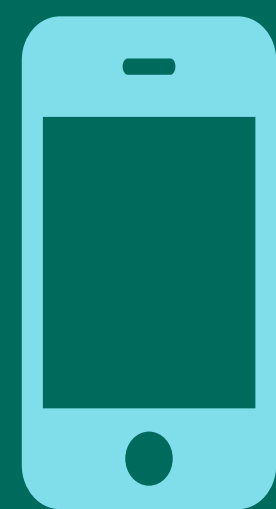
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